

SYLLABUS

CHEM 224 – ACCELERATED - Organic Chemistry B – 2nd semester
 Summer 2010 - LOYOLA UNIVERSITY CHICAGO (LUC)

Lecture/Discussion: #2322 / 2324 Sections: 003 / 004 **M+W+F: 8:30 a.m. – 11:20 a.m.** Mundelein 418

Sr. Lecturer: Dr. C. Szpunar
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 Emergency Message: Chemistry Dept. Office, 773-508-3100, fax: 773-508-3086
 Student Office Hours: M and W: after class, 11:40 – 1:00, and *** by **prior** appt ***

Required:

1. Organic Chemistry, Wade, 7th ed., Prentice Hall, 2009 (ISBN xxxxxxx)
 or Wade, 6th ed., Prentice Hall, 2003 (ISBN 0-13-147871-0)
2. Study Guide and Solutions Manual, Wade & Simek, 7th ed. (ISBN xxxx)
 or 6th ed. (ISBN 0-13-147882-6)

Suggested / Recommended Materials:

1. Molecular modeling kit, Darling, Prentice-Hall, Freeman (Maruzen), Proteus, or equivalent
2. Spiral or bound notebook for homework problems

Optional Materials (found helpful by some students):

1. Organic Chemistry as a Second Language, II, Klein (2006), Wiley (ISBN 978-0-471-73808-4)
2. Barron's Orgo Cards: Organic Chemistry Review, Wang, Razani, Lee, Wu, and Berkowitz (ISBN 0-7641-7503-3)

Grading (approx weight below) with grade guidelines: > 90%, A; 75-90%, B; 55-75%, C; grading may be curved

MIDTERM EXAM – 1 – date scheduled and announced (subject to change, although unlikely)

!!! NO MAKE UPS !!! NO EARLY EXAMS !!!

30%

- UNEXCUSED ABSENCES merit a zero score.
- EXCUSED ABSENCES are handled on a case-by-case basis; grade weighting may be adjusted, depending on the circumstance(s); however, an excused absence **MUST BE CORROBORATED and DOCUMENTED**, e.g., accompanied by a note from the doctor, dentist, hospital rep, or funeral director; by a court summons, plane ticket stub, hospital release form, obit, or other, as appropriate. With proper documentation, religious observance, official representation of the university, or personal emergency may constitute an Excused Absence.

QUIZZES – 4 – dates announced (subject to change, although unlikely) **!!! NO MAKE UPS !!!** 30%

FINAL EXAM 40%

Course Objective: To guide, encourage, and foster the learning and understanding of Organic Chemistry – nomenclature, structures, properties, reactions, mechanisms, and syntheses – by the individual student, helping him/her to connect, extrapolate, integrate, and apply the many different aspects learned.

Student Outcomes: If successful, the student will learn how to ...

1. identify the various classes of organic compounds, their methods of preparation, and typical reactions.
2. name and draw specific organic compounds.
3. postulate a logical reaction mechanism for simple organic reactions.
4. discriminate amongst relative stabilities of reaction intermediates.
5. plan and write out multi-step syntheses using known functional group transformations.
6. prepare for basic purification/separation techniques of organic compounds required in the laboratory.
7. analyze and interpret data from various instruments used in separating and identifying organic compounds: IR, NMR, and UV-vis spectrophotometers and mass spectrograph.

Lecture and Discussion – Attendance and Attention: Important and required. Feel free to bring your books and modeling kit to class. Better yet, use them. Prepare for lecture by prior scanning of new material. Come prepared for discussion, ready to ask questions on assigned homework or yet unassimilated lecture material.

Phones and Pagers: Please be courteous and respectful of others. Silent mode during lecture and discussion. **Not allowed in sight or within hearing during exams, subject to confiscation.** NO phone conversations in lecture hall or in discussion class – before class, during class, after class – AT ANY TIME!

Academic Honesty: Essential, expected, and enforced. Dishonesty dictates consequences which may include: (1) notification of Chemistry Department Chair, student's Department Chair, and CAS Dean, (2) documentation in the student's official university record, and (3) dismissal from the university. Immediate consequences will include a **ZERO** on any item in question (quiz or exam). Please refer to the LUC Undergraduate Handbook on policies or the CAS website: http://www.luc.edu/cas/pdfs/CAS_Academic_Integrity_Statement_December_07.pdf.

Study Strategies and Suggestions: One may approach the study of Organic Chemistry in a manner similar to tackling a new foreign language. Its study will provide a basis to understanding future material – building constantly, incessantly, and relentlessly on the structural and mechanistic information presented previously. Over two semesters, the course will cover functional groups, aliphatic and aromatic compounds, bonding, nomenclature, stereochemistry, conformational analysis, reaction mechanisms, multi-step syntheses, and spectroscopy. Because the course is cumulative and builds heavily on prior material, the best plan is to study Organic Chemistry regularly, every day, similar to practicing the piano. Collaboration on homework problems is encouraged, especially in a timely fashion. Experience dictates that positive outcomes (for exam and course grades) are directly proportional to working and understanding the assigned problems on a regular basis, i.e., applying the concepts learned to non-generic compounds.

Typically, Organic Chemistry is not efficiently self-taught. Overnight cramming will probably not produce success. The student should quickly read the chapter/segment to be covered BEFORE lecture to improve lecture comprehension. After lecture, careful detailed re-reading of the chapter/segment and focused working of the assigned problems are appropriate, necessary, and expected. In addition to student's participation in lecture, discussion, reading, and homework, joining and contributing to a study group is encouraged.

If anticipating a passing grade of C, the minimal time per week in the summer devoted to Organic Chemistry is estimated at 9 hr for lecture/discussion, 6-12 hr for reading, and 6-12 hr for homework.

Suggested Homework Assignment (Wade 7th edition):

- Chap 15: 1, 4-18, 22-27, 30
- Chap 16: 3-4, 7-8, 9 (a,b), 12-29, 32, 38-39, 45
- Chap 17: 1-2, 4-22, 24-27, 30-32, 38, 40-52
- Chap 18: 1-4, 6-12, 16-31, 34-40, 43-44, 47, 49, 51
- Chap 19: 1-21, 25-32, 34-42, 44, 47, 56, 58
- Chap 20: 1-33, 35-40, 45, 47, 50
- Chap 21: 1-39, 43-48, 50-54, 66
- Chap 22 : 1-47, 59-65, 67-69
- Chap 23: 1-14, 16-17, 21-22, 24-26, 28-31, 32 (a), 33-36, 40, 41 (a,b), 52-55, 63
- Chap 24: 3-6, 20, 32, 33
- Chap 25: 1, 4, 8-9, 11-15, 32
- Chap 26: FYI, 21-29

Wade 6th edition

- Chap 15: 1, 4-18, 22-27, 30
- Chap 16: 3-4, 7-8, 9 (a,b), 12-29, 32, 38-39, 45
- Chap 17: 1-2, 4-22, 24-27, 30-32, 38, 40-52
- Chap 18: 1-4, 6-12, 16-31, 34-40, 43-44, 47, 49, 51
- Chap 19: 1-21, 25-32, 34-42, 44, 47, 56, 58
- Chap 20: 1-33, 35-40, 45, 47, 50
- Chap 21: 1-39, 43-48, 50-54, 66
- Chap 22: 1-47, 59-65, 67-69
- Chap 23: 1-14, 16-17, 21-22, 24-26, 28-31, 32 (a), 33-36, 40, 41 (a,b), 52-55, 63
- Chap 24: 3-6, 20, 32, 33
- Chap 25: 1, 4, 8-9, 11-15, 32
- Chap 26: 21-29

Topics: to be covered this semester:

12/13. Spectroscopy: Methods of Structure Determination (Review)

Electromagnetic spectrum, molecular vibrations, infrared spectroscopy, characteristic IR absorptions, IR spectra interpretation, mass spectrometry (MS), terminology associated with MS, nuclear spin, magnetic shielding, NMR spectrometer, chemical shift, chemical nonequivalence, peak integration, spin-spin splitting, time dependence, interpreting NMR spectra, combined spectroscopy problems.

15. Dienes, Conjugated Systems, UV Spectroscopy

Molecular orbital theory, 1,3-butadienes, electrophilic addition (1,2 and 1,4), NBS, Diels Alder rxn.

16/17. Aromatic Compounds

Benzene, Kekule structure, resonance, annulenes, MO theory, aromaticity, Huckel's Rule, heterocyclic aromatics, electrophilic aromatic substitution, directing effects on electrophilic aromatic substitution, Friedel-Crafts alkylation and acylation, nucleophilic aromatic substitution, side-chain reactions of benzene derivatives, reactions of phenols.

18. Aldehydes and Ketones

Structure of the carbonyl group; nomenclature, physical properties, and spectroscopic properties of aldehydes and ketones; syntheses of aldehydes and ketones; addition reactions; Wittig reaction; condensation reactions; acetals; oxidation and reduction.

19. Amines

Nomenclature, structure, physical properties, basicity, salts of amines, phase transfer catalysis, spectroscopic properties, reactions of amines, sulfonamides, Hofmann elimination, Cope Elimination, arenediazonium salts, synthesis of amines.

20. Carboxylic Acids

Structure and nomenclature, physical properties, acidity, salts of carboxylic acids, spectroscopy, synthesis of carboxylic acids, nucleophilic acyl substitution, Fischer esterification, synthesis and use of acid chlorides, diazomethane, condensations of carboxylic acids, reduction, alkylation of carboxylic acids.

21. Carboxylic Acid Derivatives

Structure and nomenclature, physical properties, and spectroscopic properties of carboxylic acid derivatives, nucleophilic acyl substitution, hydrolysis of acid derivatives, reduction of acid derivatives, organometallic reactions, thioesters, carbonic acid derivatives.

22. Enols and Enolates

Enols, enolates, alpha halogenation, alkylation of enolates, formation and alkylation enamines, aldol condensation, dehydration of aldol products, crossed aldol condensation, aldol cyclizations, Claisen condensation, Dieckmann condensation, crossed Claisen condensation, β -dicarbonyl compounds, Malonic ester synthesis, acetoacetic ester synthesis, conjugate addition, Michael reaction, Robinson annulation.

23. Carbohydrates

Classification of carbohydrates, monosaccharides, erythro and threo diastereomers, epimers, nomenclature, cyclic structures of monosaccharides, anomers, mutarotation, reactions of monosaccharides, oxidation and reduction of monosaccharides, non-reducing sugars, formation of glycosides, ether and ester formation, osazone formation, Ruff degradation, Kiliani-Fischer synthesis, Fischer's proof of configuration, determination of ring size, periodic acid cleavage, disaccharides, polysaccharides.

24. Amino Acids and Peptides

Structure and stereochemistry of the α -amino acids, synthesis of amino acids, resolution, structure and nomenclature of peptides and proteins, peptide structure determination, peptide synthesis.

25. Lipids

Miscellaneous category, classification by solubility: waxes, triglycerides, soaps and detergents, phospholipids, steroids, prostaglandins, terpenes.

Lecture Outline (tentative, subject to change, but unlikely due to time constraints)

<u>Lecture</u>	<u>Date</u>	<u>Chapters</u>	<u>Topic</u>	<u>*** EVENT ***</u>
--	M-July 5	---	***** Independence Day Holiday – NO CLASS *****	
1	W-July 7	12/13	Review – IR, MS, NMR	
2	F-July 9	15	Conjugated Systems, UV Spectroscopy	
3	M-July 12	15/16	... / Aromatic Compounds	
			*****	Quiz 1 (Chapters 12/13)
4	W-July 14	16	...	
5	F-July 16	17	Aromatic Compounds - Reactions	
			*****	Quiz 2 (Chapters 15-17)
6	M-July 19	17	...	
7	W-July 21	17/18	... / Aldehydes and Ketones	
8	F-July 23	18	...	
			*****	EXAM (Chapters 12-13, 15-18)
9	M-July 26	18	...	
10	W-July 28	19	Amines	
11	F-July 30	19	...	
			*****	Quiz 3 (Chapters 18-19)
12	M-Aug 2	19/20	... / Carboxylic Acids	
13	W-Aug 4	20	...	
14	F-Aug 6	20/21	... / Carboxylic Acid Derivatives	
			*****	Quiz 4 (Chapters 20-21)
			*****	last day to withdraw with W *****
15	M-Aug 9	21/22	... / Condensations, Alpha Substitutions, Enolates	
16	W-Aug 11	23/24	Carbohydrates / Amino Acids and Peptides	
17	F-Aug 13	25	Lipids / Q/A	
			*****	Cumulative FINAL EXAM (focus: Chapters 18-23)

Daily Schedule (tentative, approximate, flexible, may adjust order):

<u>Regular Day</u>	<u>Quiz Day</u>	<u>Exam Day</u>
08:30 – 09:00 am Q/A, admin	08:30 – 09:00 am Q/A	08:30 – 09:00 Q/A
09:00 – 10:00 lecture – 1	09:00 – 10:00 lecture - 1	09:00 – 10:20 lecture
10:00 – 10:10 ***break***	10:00 – 10:10 ***break***	10:20 – 10:30 ***break***
10:10 – 10:30 discussion as time/topic permit	10:10 – 11:00 lecture – 2	10:30 – 11:20 EXAM
10:30 – 11:20 lecture - 2	11:00 – 11:20 <i>quiz</i>	*****
		08:30 – 09:10 Q/A
		09:10 – 09:20 ***break***
		09:20 – 11:20 FINAL